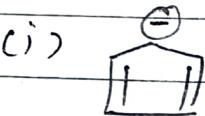
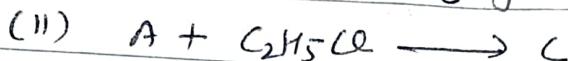
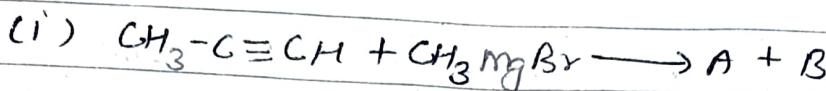


- (1) Which of the following is not electrophile  
 (a)  $\text{CH}_3^+$  (b)  $\text{SO}_3^-$  (c)  $\text{H}_3\text{O}^+$  (d)  $\text{NO}_2^+$
- (2) A prussian blue colour is formed during the test of Nitrogen from Lassaigne's extract. Blue colouration is obtained because of the formation of  
 (a)  $\text{K}_3[\text{Fe}(\text{CN})_6]$  (b)  $\text{Fe}_4[\text{Fe}(\text{CN})_6]_3$   
 (c)  $\text{Fe}_3[\text{Fe}(\text{CN})_6]_2$  (d)  $\text{Fe}_4[\text{Fe}(\text{CN})_5\text{NO}_2]_3$
- (3) Which of the following species is not aromatic  
 (i)  (ii)  (iii)   
 (iv) 
- (4) Which of the following will give a positive Fehling's test  
 (i) Benzaldehyde (ii) Fructose  
 (iii) Acetophenone (iv) Salicylic acid.
- (5) Which of the given structure represent Neoprene  
 (i)  $\left\{ \text{CH}_2-\text{CH}=\overset{\text{CH}_3}{\underset{\text{C}}{\text{C}}}-\text{CH}_2 \right\}_n$  (ii)  $\left\{ \text{CH}_2-\text{CH}=\overset{\text{Cl}}{\underset{\text{C}}{\text{C}}}-\text{CH}_2 \right\}_n$   
 (iii)  $\left\{ \overset{\text{CH}_3}{\underset{\text{C}}{\text{C}}}-\text{CH}_2 \right\}_n$  (iv)  $\left\{ \text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2 \right\}_n$

(6)

Consider the following sequence of reaction



$C$  is :-

(i) Propane

(iii)  $\text{CH}_3\text{C}\equiv\text{CCH}_3$

(ii)  $\text{CH}_3\text{C}\equiv\text{CCH}_3$

(iv) Butane.

(7)

Baeyer reagent test is given by

(a) Acetophenone

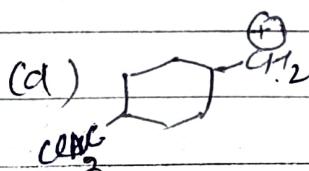
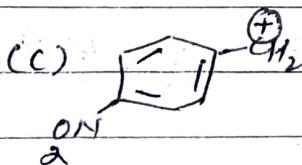
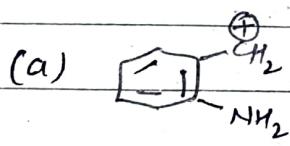
(b) Anthracene

(c) Acetone

(d) Cinnamaldehyde

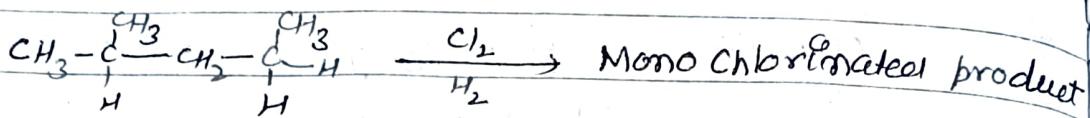
(8)

Amongst the following the most stable carbocation is



(9)

Consider the following reaction



How many optically active compounds are produced in this reaction?

- (a) 2
- (b) 4
- (c) 6
- (d) 8

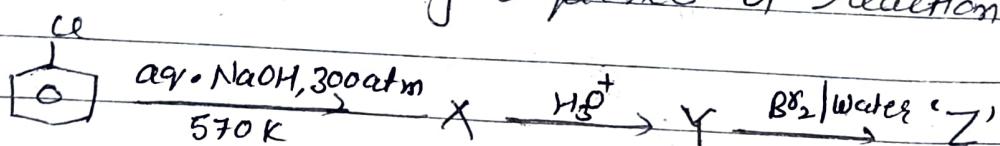
(10)

The Hinsberg test for amines, the reaction type is:-

- (i) Nucleophilic substitution rxn
- (ii) Electrophilic Substitution reaction
- (iii) Elimination reaction
- (iv) Rearrangement reaction.

(11)

Consider the following sequence of reaction.

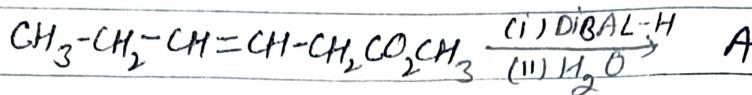


Z is

- (a)  $\text{Ce}(\text{OH})_3\text{Br}$
- (b)  $\text{Ce}(\text{OH})_2\text{Br}_2$
- (c)  $\text{Ce}(\text{OH})\text{Br}_3$
- (d)  $\text{Ce}(\text{Br})_3\text{OH}$

(12)

In the given reaction

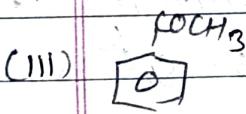
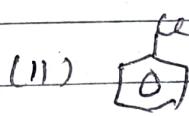
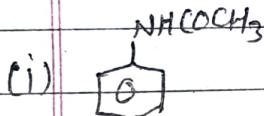


A is

- (a)  $\text{CH}_3-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{OH}$
- (b)  $\text{CH}_3-\text{CH}_2-\text{CH}=\text{CH}-\text{CH}_2-\text{CH}_2-\text{OH}$
- (c)  $\text{CH}_3-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{OH}$
- (d)  $\text{CH}=\text{CH CH}_2\text{CH}_2\text{CH}_2\text{CHO}$

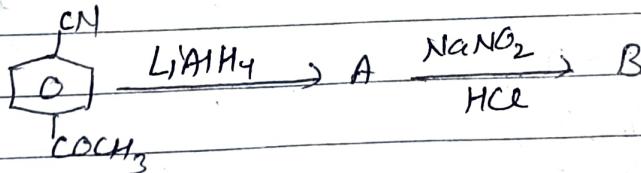
(13)

Which of the following compounds form the meta substitution product as the major product

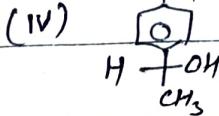
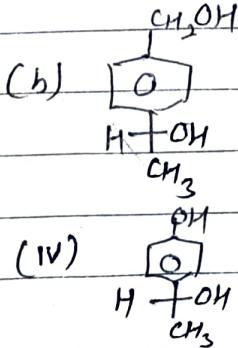
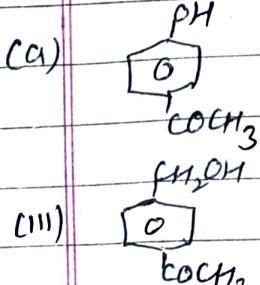


(14)

In the given sequence of reaction



B is



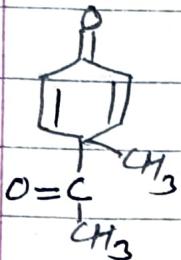
15

The species which is least stabilised by resonance is

(I)



(II)



(III)



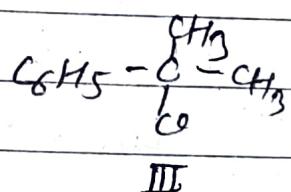
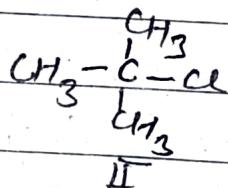
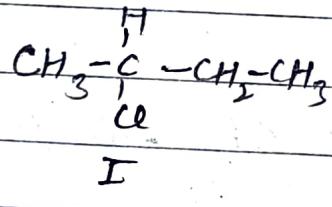
(IV)



16

The decreasing order of reactivity of the following halides towards  $\text{Sn}^+$  is:-

(Ea)



(I)

$$\text{I} > \text{III} > \text{II}$$

(II)

$$\text{II} > \text{III} > \text{I}$$

(III)

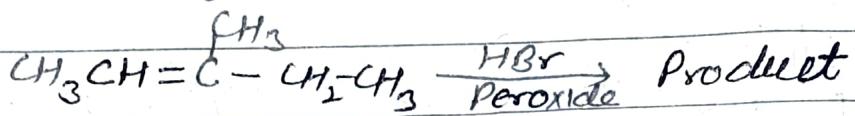
$$\text{III} > \text{II} > \text{I}$$

(IV)

$$\text{II} = \text{III} > \text{I}$$

(17)

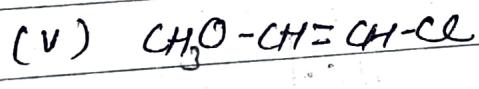
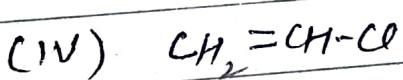
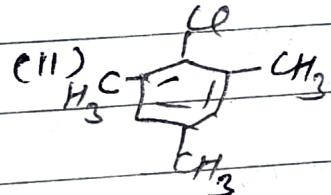
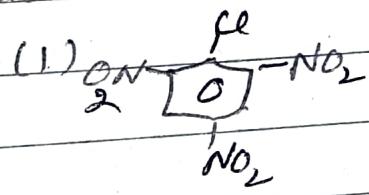
In the given reaction



The number of possible stereoisomerism for the product is/are

- (I) Two
- (II) four
- (III) zero
- (IV) Six

(18) Which of these species will undergo Nucleophilic Substitution reaction most readily



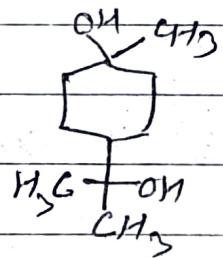
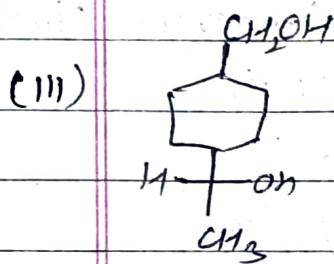
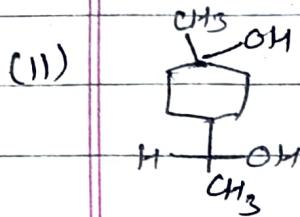
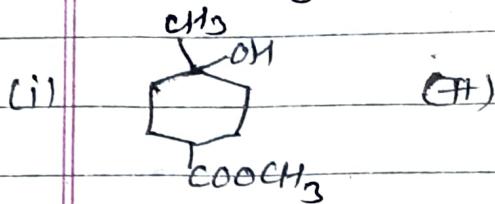
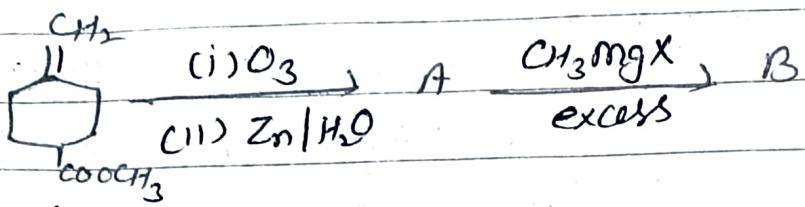
(19)

Chloroform is always kept in dark brown bottles. It reacts with the oxygen present in air to form

- (i)  $\text{Cl}_2$
- (ii)  $\text{HCOOH}$
- (iii)  $\text{HCHO}$
- (iv)  $\text{COCl}_2$

(20)

On the given sequence of reaction



(21)

The number of possible structural isomerism of an alkyne with molecular formula  $\text{C}_5\text{H}_8$  is

- (a) Two
- (b) Three
- (c) four
- (d) five